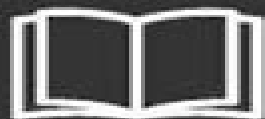


Travis Maynard

# Getting Started with Gulp

**Second Edition**

Create powerful automations with Gulp to improve the efficiency of your web project workflow



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# Getting Started With Gulp Maynard Travis

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the  $\text{h}^+$  ions from the acid and the  $\text{oh}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm<sup>3</sup> place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol<sup>-1</sup>

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eg 1  $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o l}$  for which  $\Delta h = 57\text{ kJ mol}^{-1}$  eg 2  $\text{h}_2\text{so}_4 \text{ aq} + \text{koh aq} \rightarrow \text{ks}_2\text{o}_7 \text{ aq} + \text{h}_2\text{o l}$  for which  $\Delta h = 57\text{ kJ mol}^{-1}$

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