#### Chemistry Semester 1 Final Exam Review

 What is a definition of Chemistry? Study of matter Matter includes all of the following EXCEPT: by Birchit c. smoke a. air d. water vapor What is manter? Something that has mass and takes up space (volume) 4. A physical change occurs when a at peach speaks e. copper bowl tarnishes b. bracelet turns your wrist green d. glue gun melts a glue stick 5. The particles in a solid are a, packed closely together c. constantly in motion b. viery far apart d, able to slide past one another Of the following groups, 1, 2, 14, 15, 17, 18: Group(s) 1,2 contains only metals Group (s) 18 is called a Noble gases 8. Group(s) 17 has 7 electrons on the valence shell 9. Group(s) 15 needs to gain 3 electrons to fill a stable octet 10. Group(s) 2 has a 2+ exidation state has an electron dot notation of .X. Group(s) 2 Using the periodic table, an element similar to carbon would be Based on their location on the periodic table, you could infer that 18/noble gases are very unreactive The horizontal row on the periodic table is called a(n) period The vertical column of blocks on the periodic table is called a(n) group/family The elements on the zigzag line in the periodic table are metalloids 17. Which is NOT a property of metal? a. malleability e, unreactivity b, ability to lose electrons d. ability to conduct heat. A volume of I cc (cm<sup>2</sup>) is the same as To determine density, the quantities that must be measured are volume mend manage The relationship between the mass m of a material, its volume V, and its density D is a, V = snDb.  $F_{\text{int}} = D$  $c_{-}DV = m$  $d:D+V=\infty$ 28. The density of aluminum is 2.70 g/cm3. The volume of a solid piece of Aluminum is 1.50 cm3. Find the mass 4.05 grams The density of pure diamond is 3.5 g/cm<sup>3</sup>. The mass of a diamond is 0.25 g. Find the volume = 0.71 cm<sup>3</sup>. 22. What statement about density is true? a. two samples of a pure substance may have different densities if they are different sizes the density of a sample depends on its location on Earth c. a cylinder is always used to measure the volume d. density is a physical property that remains constant for a pure substance The number of grams equal to 0.5 kg is 500 g In a graph the independent variable is placed on the x axis and the dependent variable is placed on the y axis. The variable that you change in the lab is the Independent variable. The variable that responds to the change is called the dependant variable. All other things do not change and are called constants At first Dalton thought the atom was solid, spherical, indivisible , then Ruther ford discovered the mucleus of the atom, then the subatomic particles protons , neutrons , and electrons were discovered 27. The positively charged particle is the proton found in the nucleus of the atom. 28. The subatomic particle that has about the same mass as the proton, but with no electrical charge, is called a and is found in the nucleus of the atom 29. The electron is negatively charged and is found in the electron cloud outer shell of the atom. 30. Bhor is the person credited with placing electrons on levels and Schrodinger established the electron cloud theory. 31. The forces that hold the particles together in the nucleus are columbic /electrostatic Concrete. 32. Isotopes are atoms of the same element that have different number of neutrons 34. What is the atomic number for Nitrogen? 7 To what subatomic particle does this correspond? # of protons 35. Zinc -66 has 30 protons and 36 mourtrons 36. What is the electron configuration for sodium? [Ne]3s\*Chlorine? [Ne] 3s\*3p\* Neon? [He] 2s\*2p\*

37. What is the dot notation for magnesium? Bromine? Argon?

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