

Name _____ Class _____ Date _____

- 1 Write each of the changes listed below in the correct column in the table to show whether they are physical changes or chemical reactions.

Changes: boiling burning melting neutralising

Physical changes	Chemical reactions

- 2 Some of the statements below are true and some are false.

Put a tick (✓) in the box next to the statements that are true.

- ☐ New substances are made during a chemical reaction.
- ☐ New substances are made during a physical change.
- ☐ A chemical reaction is usually easier to reverse than a physical change.
- ☐ A physical change is usually easier to reverse than a chemical reaction.
- ☐ You cannot reverse physical changes or chemical reactions.

- 3 Each observation below is describing a physical change or a chemical reaction.

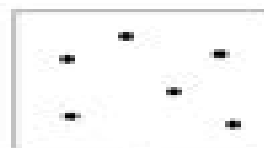
Put a tick (✓) in the correct column to show whether each is a physical change or a chemical reaction.

Observation	Physical change	Chemical reaction
A liquid was set on fire and it burnt with a yellow flame.		
A liquid was cooled down and it changed into a solid.		

- 4 The boxes A and B show gas particles in containers.



A



B

Which box shows the gas at the higher pressure? _____

Explain your answer. _____

I can...

- identify and explain the differences between physical changes and chemical reactions
- use particle theory to explain gas pressure and how it can be changed.

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Table of Contents Exploring Science 9i Summary Sheets

1. Understanding the eBook Exploring Science 9i Summary Sheets
 - The Rise of Digital Reading Exploring Science 9i Summary Sheets
 - Advantages of eBooks Over Traditional Books
2. Identifying Exploring Science 9i Summary Sheets
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exploring Science 9i Summary Sheets
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exploring Science 9i Summary Sheets
 - Personalized Recommendations
 - Exploring Science 9i Summary Sheets User Reviews and Ratings
 - Exploring Science 9i Summary Sheets and Bestseller Lists
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- Exploring Science 9i Summary Sheets eBook Subscription Services
- Exploring Science 9i Summary Sheets Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Exploring Science 9i Summary Sheets Compatibility with Devices
 - Exploring Science 9i Summary Sheets Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exploring Science 9i Summary Sheets
 - Highlighting and Note-Taking Exploring Science 9i Summary Sheets
 - Interactive Elements Exploring Science 9i Summary Sheets
- 8. Staying Engaged with Exploring Science 9i Summary Sheets
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exploring Science 9i Summary Sheets
- 9. Balancing eBooks and Physical Books Exploring Science 9i Summary Sheets
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exploring Science 9i Summary Sheets
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Exploring Science 9i Summary Sheets
 - Setting Reading Goals Exploring Science 9i Summary Sheets
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Exploring Science 9i Summary Sheets
 - Fact-Checking eBook Content of Exploring Science 9i Summary Sheets
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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