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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Explore Learning Ionic Bonds Answers

Sylvia S. Mader, Michael Windelspecht



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The Art and Science of Questions Ronald Legarski, 2024-09-28 In *The Art and Science of Questions* Ronald Legarski explores the profound power of questioning in shaping human thought innovation and discovery Whether in business science education or everyday life the ability to ask the right question can lead to deeper understanding better decisions and groundbreaking ideas This book offers a comprehensive guide to mastering the art of inquiry covering various types of questions from open ended to leading reflective and hypothetical and providing practical methods for applying them in real world contexts With insights drawn from philosophy psychology leadership and technology this book equips readers with the tools to ask more effective questions and unlock the full potential of inquiry Legarski also delves into the future of questioning examining how artificial intelligence virtual reality and quantum computing will revolutionize how we ask and answer questions in the coming decades Whether you re a student a professional or simply a curious mind *The Art and Science of Questions* will transform the way you approach problem solving learning and communication

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Investigating Chemistry Matthew Johll, 2008-12-22 In its new second edition *Investigating Chemistry A Forensic Science Perspective* remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package

Reading Actively in Middle Grade Science Don K. Philpot, 2020-09-15 Middle grade students can learn a great deal about themselves and their world by reading informative texts in science courses These texts will focus on important topics in earth science life science and physical science and shape students understandings about scientific inquiry science related processes and phenomenon engineering and design and technological innovations But reading is a complex act and most students need specific reading related support to understand assigned texts in middle grade science courses This book focuses on the cyclical nature of reading the actions proficient readers engage in to understand science textbooks and other informational texts and the instructional support that teachers can provide to enhance middle grade students learning of science content through reading Three associated questions will be addressed in this book What actions do proficient readers engage in to understand assigned course texts

What do these actions entail and how do they relate to each other What teacher mediated practices best support middle grade students development as proficient readers and enhance their learning of course content through reading

An Introduction to Physical Science James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between

Biology Sylvia S. Mader, Michael Windelspecht, 2021 Biology Fourteenth edition is an understanding of biological concepts and a working knowledge of the scientific process

Teach Now! Science Tom Sherrington, 2014-05-09 Being taught by a great teacher is one of the great privileges of life Teach Now is an exciting new series that opens up the secrets of great teachers and step by step helps trainees to build the skills and confidence they need to become first rate classroom practitioners Written by a highly skilled practitioner this practical classroom focused guide contains all the support you need to become a great science teacher Combining a grounded modern rationale for learning and teaching with highly practical training approaches the book guides you through all the different aspects of science teaching offering clear straightforward advice on classroom practice lesson planning and working in schools Teaching and learning planning assessment and behaviour management are all covered in detail with a host of carefully chosen examples used to demonstrate good practice There are also chapters on organising practical work the science curriculum key ideas that underpin science as a subject and finding the right job Throughout the book there is a wide selection of ready to use activities strategies and techniques to help you bring science alive in all three main disciplines including common experiments and demonstrations from biology physics and chemistry to engage and inspire you and your students Celebrating the whole process of engaging young people with the awe and wonder of science this book is your essential guide as you start your exciting and rewarding career as an outstanding science teacher

Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04 Matthew Johll s Exploring Chemistry covers the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book s specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2000 Zumdahl s conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts

before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations The Software Encyclopedia 2001 ,2001 **Energy Research Abstracts** ,1990 Science News-letter ,1963 Atoms & Chemical Bonding Science Learning Guide NewPath Learning,2014-03-01 The Atoms Atomic Configuration Chemical Bonding Ionic Bonding Ionic Compounds Covalent Bonding Covalent Compounds Naming Compounds and Metallic Bonding Aligned to Next Generation Science Standards NGSS and other state standards

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