Substance	Lewis Structure	Molar Mass	Intermolecular Attractive Force
Methanol CH <sub>3</sub> OH 4+3(1)+6+1+34	H-C-Ö+H	32.042	Hydrogen
Ethanol CH <sub>3</sub> CH <sub>2</sub> OH	H-C-C-O-H	46.068	Hydrogen
n-Butanol CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OH	H-0-0-1 H-0-0-1 H-0-0-1	74.122	Hydrogen
Glycerin C <sub>3</sub> H <sub>s</sub> (OH) <sub>3</sub>	H-C-C-C-H 19-HH 19-H	90.093	Hydrogen
Water H <sub>2</sub> O	H-0-H	18.015	Hydrogen Bonds

# **Exploring Intermolecular Forces Lab Answers**

**Yijin Wang** 

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